## Brèves communications - Kurze Mitteilungen - Brevi comunicazioni - Brief Reports

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## Discontinuous Nature of Stress-Strain Relationship

Two 0.0138 in. dia. piano wires (E =  $28.65 \times 10^6$  lb/in<sup>2</sup> at 200,000 lb/in2, ultimate stress = 320,000 lb/in2) are prestressed, then clamped with specially hardened and serrated grips in a stressing bed 2 feet long, made of  $4 \times 2$  in. channel sections. While one wire is sufficient for the experiment, the second one served for measuring the relaxation of the stress. If a small weight is placed at the middle of the span the wire will deflect assuming a triangular shape. The major load, causing a deflection,  $\delta \approx 20 \times 10^{-3}$ in., was followed by minor loads of 100, 50, 20 and 10 mg. An increase in  $\delta$  was observed upon applying the first two minor riders, but none for 20 or 10 mg. This meant that an increase in the wire stress was not followed by an appropriate increase in strain. The calculated deflection for a 20 mg increment was  $9.3 \times 10^{-6}$  in, which, due to the geometry of the wire (25.92 in. long), meant  $10 \times 10^{-10}$ strain with a corresponding stress of  $28 \times 10^{-3}$  lb/in<sup>2</sup>. This stress, which is of the order of 0.1 lb/in2, will be referred to as the entry stress and is defined as the tension below which stress is not followed by strain.

The basic difficulty in finding the entry stress lies in the limited accuracy of micrometers for measuring travel. A large drum micrometer with  $10^{-4}$  in. graduations was used, with an accuracy of  $5 \times 10^{-5}$  in. The electronic contact indicator will register a maximum gap between wire and micrometer shaft of the order of  $10^{-6}$  in. but the exact magnitude of this gap is impossible to find by a mechanical measuring device and recourse has to be made to the differential thermal expansion that exists between the wire and the stressing bed.

Let  $\varepsilon_{\rm s}$  and  $\varepsilon_{w}$  be these expansion coefficients of the steel frame and of the wire respectively and let  $\varepsilon_{w} = \varepsilon_{\rm s} + \Delta \varepsilon$ . It may be shown that the following relationship exists between  $\varepsilon_{\rm s}$  and  $\Delta \varepsilon$ :

$$\varphi \Delta \varepsilon \left(1 - \varphi \Delta \varepsilon\right) + \frac{\alpha^2}{2} = \frac{\delta_{\varphi}^2}{2 \, l^2} \left(1 - 2 \, \varphi \, \varepsilon_{\rm s}\right) \tag{1}$$

where  $\varphi$  is the temperature change,  $\alpha$  is the angle between the horizontal and the slope of the wire before  $\varphi$  takes place, and  $\delta_{\varphi}$  is the deflection of the centre of the span after the temperature increase. From equation 1  $\varDelta \varepsilon$  can be found if  $\varepsilon_{\rm s}$  is known; however, it may be shown that even if the large error of  $1\times 10^{-6}$  in  $\varepsilon_{\rm s}$  is assumed the error in  $\varDelta \varepsilon$  is of the order of  $10^{-10}$ , hence  $\varepsilon_{\rm s}=6\times 10^{-6}$  was adopted for the mild steel frame.

Having found  $\Delta \epsilon$  to be  $4.4 \times 10^{-7}$  per F the exact magnitude of the sensitivity gap between wire and micrometer

need not be known as contact can be achieved by lifting the wire slowly during decreasing temperature. It may be shown that

$$d \delta_{\varphi}/dt = \Delta \varepsilon (l^2/\delta_{\varphi}) d\varphi/dt$$
 (2)

where the left-hand derivative is the rate of the rise of the wire, while the right-hand derivative is the rate of change of the temperature (cooling is positive).

The cooling down process took place in a windowless room, free of draught, during the evening hours when the temperature decrease was fairly uniform. Three experiments have been carried out on different days when the temperature drop was 0.8 F per hour and the speed of the ascending wire was calculated as  $5.5 \times 10^{-5}$  in./min.

When, now, while the wire was ascending and the contact indicator arc began to vibrate, a load of 20 mg was added to the load the contact should have immediately been broken because this load increment, as mentioned before, should have caused a deflection of  $9.3 \times 10^{-6}$  in., a distance the wire takes more than 10 sec to travel. Failure of breaking the contact means that 20 mg was insufficient to cause a change of strain.

This result was obtained for  $\sigma \approx 200,000 \text{ lb/in}^2$  prestress (less relaxation). Similar experiments were carried out for 65,000, 100,000 and 150,000 lb/in² prestress with no significant change in the entry stress. The ratio of the stresses in the wire as against those in the stressing bed and contacts is of the order of 104. It follows that effects due to the entry stress are restricted solely to the wire.

It is emphasized that 200,000 lb/in² stress is not high enough for the load to be carried by the skin as against the core. Hence, even if crystals begin to slip dislocations moving towards the free surface will not have to travel through many atomic spacings as this would involve an appreciable plastic strain in the core. The dislocation theory cannot, therefore, explain the entry stress.

Zusammenfassung. Wird ein vorgespannter, dünner Draht mit zunehmender Kraft in der Mitte belastet, so ist jeder Lastzuwachs mit einer Zunahme der Durchbiegung verbunden. Dies trifft nicht mehr zu, falls der Lastzuwachs einen gewissen Wert unterschreitet, was bedeuten würde, dass für kleine Spannungsänderungen unterhalb der Grössenordnung 0,1 lb/in² das Hooksche Gesetz ungültig ist.

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## Preparation of 5,6-o-Isopropylidene-L-ascorbic Acid<sup>1</sup>

It has been found that, while L-ascorbic acid is sparingly soluble in acetone, it is very soluble in acetone saturated with hydrogen chloride. However, 5,6-o-iso-propylidene-L-ascorbic acid exhibits the opposite characteristics, being poorly soluble in acetone saturated with hydrogen chloride and readily soluble in acetone. It is

possible to take advantage of these solubility properties to prepare the isopropylidene derivative in high yield. The condensation can also be accomplished in lower yields by shaking with anhydrous copper sulfate for 24 h or treatment with sulfuric acid overnight<sup>2</sup>. The product is

<sup>&</sup>lt;sup>1</sup> Supported by PHS Research Grant CA-03994-06.

<sup>&</sup>lt;sup>2</sup> L. v. Vargha, Nature 130, 847 (1932).